

Problem on Source partitioning

There are some very helpful programs available on the web from Don Phillips at the EPA for source partitioning by both statistical and quantitative methods. There is also an R package “simmr” that does the same thing.

You can find the simple spreadsheet “Isoerror” that you can use – we will put this on the course web site. For Problem 2a you will need to use one of the two programs below that give probability distributions when you have multiple sources.

Please download one of these and for answering the following questions:

Isosource <http://www.epa.gov/wed/pages/models/stableIsotopes/isotopes.htm>

Simmr <https://cran.r-project.org/web/packages/simmr/vignettes/simmr.html>

If using the Simmr, you will need R (also recommended Rstudio), and

a) You are trying to detect the contribution of heterotrophic decomposition to total ecosystem respiration fluxes, and have measured the respiration $\Delta^{14}\text{CO}_2$ value to be +118‰ (SD±10, n=4). From your lab incubations of soil and plant respiration at your site, you found the $\Delta^{14}\text{CO}_2$ values to be +145‰ (SD±20, n=2), and +75‰ (SD±5, n=6) respectively. Please calculate the proportional contribution of heterotrophic respiration using IsoError. What would happen to your estimate if you had more replicates of your measurements? What happens if the inherent variability of your observations was larger, or smaller?

(for this you can use the Isoerror spreadsheet, on the web site).

DO PARTS B and C only if you have the on-line Isoerror or simmr....

b) Now you want to understand how the surface and deep soil each contribute individually to the total ecosystem respiration value, so you make measurements in another site and find the respiration $\Delta^{14}\text{CO}_2$ value to be +31‰ (SD±8, n=4). From your lab incubations of surface soil, deep soil, and plant respiration at your site, you found the $\Delta^{14}\text{CO}_2$ values to be +98‰ (SD±15, n=3), -13‰ (SD±15, n=3), and +65‰ (SD±5, n=4) respectively. Please calculate the statistical proportional contribution of the three components to total ecosystem respiration using IsoSource or simmr (you can assume $\delta^{13}\text{C}$ of all observations was -25 per mil).

c) Because you are left with an unsatisfied craving after your statistical partitioning, you decide to reanalyze your data by including $\delta^{13}\text{CO}_2$ data that you also collected from your sites. Your measurement of ecosystem respiration $\delta^{13}\text{CO}_2$ was -24.6‰ (SD±0.2, n=3). From your lab incubations of surface soil, deep soil, and plant respiration at your site, you found the $\delta^{13}\text{CO}_2$ values to be -26.2‰ (SD±0.3, n=3), -23.7‰ (SD±0.2, n=3), and -23.5‰ (SD±0.4, n=3) respectively. Please combine these measurements with those of 2b and calculate the contribution of the three components to ecosystem respiration using IsoError or simmr.

Problem on Fossil fuel emissions

On the web site you will find an excel file with data taken at Caltech in the Los Angeles basin. This basin has a strong night-time air inversion that traps air, so concentrations of CO₂ rise at night. Using either the mixing approach or the Keeling plot approach (see below), estimate how much fossil fuels contribute to night-time CO₂ increase in the LA basin on the nights given.

Isotope mixing – linear superposition of two sources
(otherwise known as the “Keeling plot”)

$C_m = C_{bg} + C_{ff}$ mass balance for sum of isotopes where C = concentration in mixture (m), and sources bg (background atmosphere) and ff (fossil fuel)

$^{13}C_m = ^{13}C_{bg} + ^{13}C_{ff}$ mass balance for isotope (¹³C or ¹⁴C)

$\delta_m C_m = \delta_{bg} C_{bg} + \delta_{ff} C_{ff}$ \approx mass balance for isotope ($\delta^{13}C$ or $\Delta^{14}C$)

with some algebra (see the papers referenced below for the derivation)

$$\delta_m = \underbrace{(\delta_{bg} C_{bg} - \delta_{ff} C_{bg})}_{\text{slope}} \cdot (1/C_m) + \underbrace{\delta_{ff}}_{\text{intercept}}$$

Keeling plot assumptions:

- 1) C_A , d_A and d_B are unchanging
- 2) mixing is mass-independent (turbulent, no diffusion)
- 3) no chemical or equilibrium reactions
- 4) only 2 sources A and B mix
(above are usually met for respiration at night)
- 5) OR if there is a third source (C) then the relative proportions of B and C do not change, so their combined isotope ratio is the same

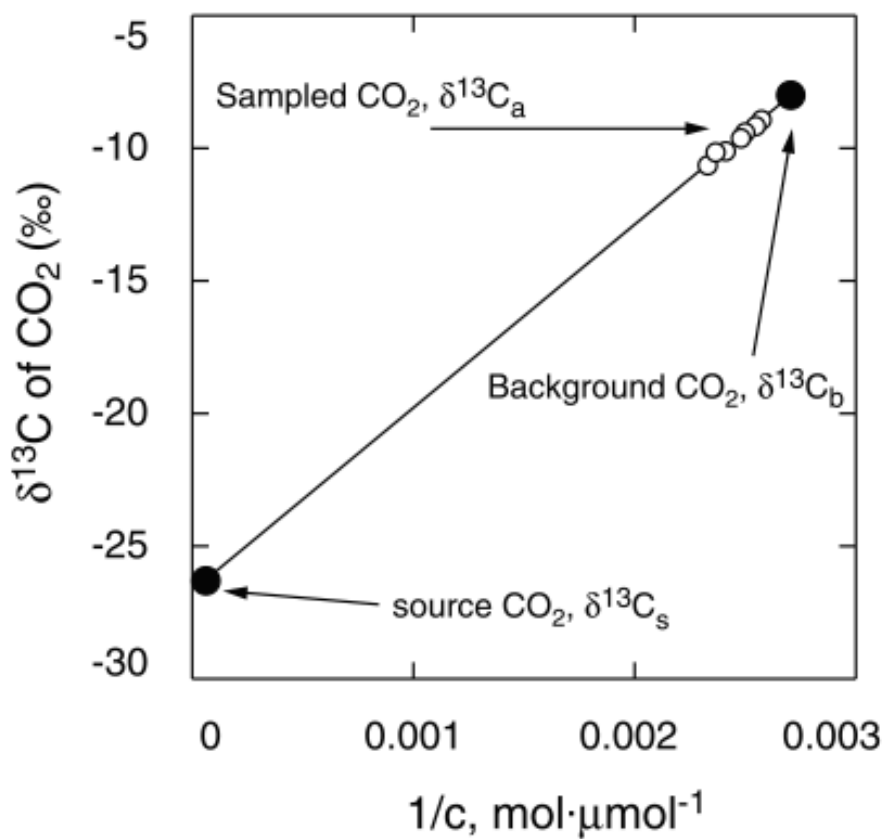


Figure 1. Graphical illustration of the Keeling plot method given as equation (3) in the text. The carbon isotope composition of two endpoints of source CO₂ ($\delta^{13}\text{C}_s$) and well-mixed, background atmospheric CO₂ ($\delta^{13}\text{C}_b$) are shown in solid circles. The carbon isotope composition of sampled air ($\delta^{13}\text{C}_a$) is shown by open circles. Isotope ratios are plotted against the inverse of CO₂ concentration (c). Note the distance of the samples from the intercept.

Keeling, C.D. The concentration and isotopic abundances of atmospheric carbon dioxide in rural areas. *Geochimica et Cosmochimica Acta* 13: 322-334, 1958.

Pataki, D. E.; Ehleringer, J. R.; Flanagan, L. B.; Yakir, D.; Bowling, D. R.; Still, C. J.; Buchmann, N.; Kaplan, J. O.; Berry, J. A. The application and interpretation of Keeling plots in terrestrial carbon cycle research *Global Biogeochem. Cycles* Vol. 17 No. 1 doi 10.1029/2001GB001850